

Detection of ^{63}Ni and ^{59}Ni by AMS: Applications in Hiroshima Dosimetry and Biomedical Tracing.[†]

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The reaction $^{63}\text{Cu}(n,p)^{63}\text{Ni}$ has been identified as one of a small number of reactions which might be used for the direct determination of the Hiroshima fast neutron fluence (Marchetti and Straume, 1995, Appl. Radiat. Isotop. ??). We have developed chemical techniques for the separation of ^{63}Ni ($t_{1/2} = 100$ y) from bulk copper, followed by measurement of ^{63}Ni by accelerator mass spectrometry. Electrochemical separation is used to isolate ^{63}Ni and trace stable Ni from gram-sized copper samples. Stable Ni (1 mg) is added as a carrier. Reaction with carbon monoxide to form $\text{Ni}(\text{CO})_4$ is used to further reduce Cu to $< 2 \times 10^{-8}$ (Cu/Ni). The $\text{Ni}(\text{CO})_4$ is thermally decomposed directly in sample holders for measurement by AMS. After analysis in the AMS spectrometer, the ions are identified using characteristic projectile x-rays. This allows the further rejection of the remaining ^{63}Cu , which is a stable atomic isobar of ^{63}Ni . In a demonstration experiment, ^{63}Ni was measured in Cu wires (2-20 g) which had been exposed to a ^{252}Cf source. We successfully measured ^{63}Ni at levels necessary for the measurement of Cu samples exposed near the Hiroshima hypocenter. For the demonstration samples, the Cu was chemically reduced by a factor of 10^{12} with quantitative retention of ^{63}Ni . Detection sensitivity (3 σ) was ~ 20 fg ^{63}Ni ($^{63}\text{Ni}/\text{Ni}$ (2×10^{-11})). A significant improvement in sensitivity is expected with incremental changes in the methods. Preliminary results indicate that a similar sensitivity is achievable for ^{59}Ni ($t_{1/2} = 105$ y). We will report on initial work on the application of this isotope as a biomedical tracer in living systems.

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